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5-Phenoxymethyl-1,3,4-oxadiazole-2(3H)-thione

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Thamotharan *et al.* • $C_9H_8N_2O_2S$

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Key indicators

Single-crystal X-ray study T = 160 KMean σ (C–C) = 0.006 Å R factor = 0.047 wR factor = 0.105 Data-to-parameter ratio = 11.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

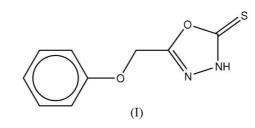
5-Phenoxymethyl-1,3,4-oxadiazole-2(3H)-thione

In the title compound, $C_9H_8N_2O_2S$, the H atom of the thiol group has been transferred to the neighbouring N atom of the oxadiazole ring. Intermolecular $N-H\cdots N$ hydrogen bonds exist between adjacent molecules.

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Comment

It is well known that 1,3,4-oxadiazole-2-thione derivatives show a broad spectrum of biological activities (Ram & Vlietinck, 1988; Boschelli *et al.*, 1993). A view of the title compound, (I), with the atomic numbering scheme, is shown in Fig. 1. The dihedral angle between the mean planes of the benzene and 1,3,4-oxadiazole rings is 14.4 (1)°. In (I), the bond lengths and angles are in good agreement with the expected values (Allen *et al.*, 1987). The H atom of the thiol group has been transferred to the adjacent N atom of the oxadiazole ring. The N3–N4 [1.383 (5) Å] and C2=S2 [1.647 (4) Å] bond lengths correspond to the usual single N–N and double C=S distances.



The crystal structure of (I) is shown in Fig. 2. In the solid state, atom N3 is involved in an intermolecular $N-H\cdots N$ hydrogen bond with atom N4 of the oxadiazole group of an adjacent molecule (Table 1). This hydrogen bond links the molecules into chains, which run parallel to the [010] direction and can be described by a C(3) graph-set motif (Bernstein *et al.*, 1995). Atom C6 (*via* H61) acts as a donor for a weak

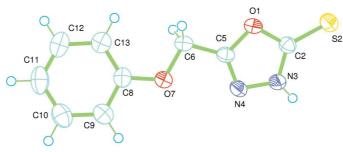


Figure 1

A view of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are shown as small spheres of arbitrary radii.

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intermolecular C–H···O interaction with atom O1 of a symmetry-related molecule. This weak interaction connects the molecules into chains, which also run parallel to the [010] direction and can be described by a graph-set motif of C(4). In addition, atom C6 (*via* H62) is involved in an intermolecular C–H··· π interaction with the benzene ring of a neighbouring molecule [H62···Cg = 2.71 Å, C6···Cg = 3.459 (4) Å and C6–H61··· $Cg = 132^{\circ}$, where Cg is the centroid of the benzene ring at (x, y - 1, z)].

Experimental

A solution of phenoxyacetic acid hydrazide (0.01 mol) was dissolved in pyridine (10 ml), and carbon disulfide (5 ml) was added with constant stirring. Stirring was continued for 36 h at room temperature. The reaction mixture was then poured into ice-cold water and acidified with dilute HCl. The solid, (I), separated, was filtered off and crystallized from dimethylformamide (m.p. 457–459 K).

1553 independent reflections

 $\begin{aligned} R_{\rm int} &= 0.083\\ \theta_{\rm max} &= 25.0^\circ \end{aligned}$

 $h = -11 \rightarrow 11$

 $k = -5 \rightarrow 5$

 $l = -11 \rightarrow 12$

1295 reflections with $I > 2\sigma(I)$

Crystal data

$C_9H_8N_2O_2S$	$D_x = 1.470 \text{ Mg m}^{-3}$
$M_r = 208.23$	Mo $K\alpha$ radiation
Monoclinic, P2 ₁	Cell parameters from 21914
a = 9.3233 (8) Å	reflections
b = 4.9446 (5) Å	$\theta = 2.0-25.0^{\circ}$
c = 10.2051 (10) Å	$\mu = 0.32 \text{ mm}^{-1}$
$\beta = 91.388 \ (5)^{\circ}$	T = 160 (2) K
V = 470.32 (8) Å ³	Needle, colourless
Z = 2	$0.35 \times 0.10 \times 0.02~\text{mm}$

Data collection

Nonius KappaCCD area-detector
diffractometer
φ and ω scans with κ offsets
Absorption correction: multi-scan
(SORTAV; Blessing, 1995)
$T_{\min} = 0.676, T_{\max} = 0.999$
6143 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0334P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.047$	+ 0.2737P]
$wR(F^2) = 0.105$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.08	$(\Delta/\sigma)_{\rm max} < 0.001$
1553 reflections	$\Delta \rho_{\rm max} = 0.19 \ {\rm e} \ {\rm \AA}^{-3}$
131 parameters	$\Delta \rho_{\rm min} = -0.19 \text{ e} \text{ \AA}^{-3}$
H atoms treated by a mixture of	Absolute structure: Flack &
independent and constrained	Bernardinelli (1999, 2000), 636
refinement	Friedel pairs
	Flack parameter: 0.01 (14)

Ta	ble	1	
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Hydrogen-bond	geometry	(À,	°).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$\begin{matrix} N3-H3\cdots N4^i\\ C6-H61\cdots O1^{ii} \end{matrix}$	0.87 (5)	2.24 (4)	2.899 (5)	132 (3)
	0.99	2.53	3.416 (4)	148

Symmetry codes: (i) -x + 1, $y - \frac{1}{2}$, -z; (ii) -x, $y + \frac{1}{2}$, -z.

The position of the amine H atom was determined from a difference Fourier map and refined freely along with its isotropic displacement parameter. All remaining H atoms were placed in geometrically idealized positions and were constrained to ride on

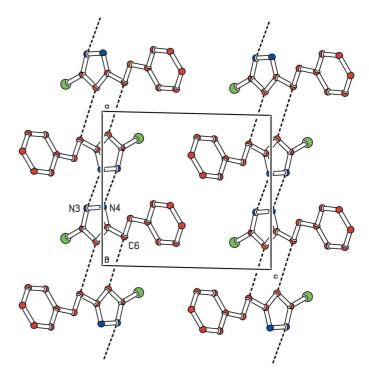


Figure 2

Crystal structure of (I), as projected on to the *ac* plane. $N-H\cdots N$ and $C-H\cdots O$ bonds are indicated by dashed lines. H atoms have been omitted.

their parent atoms, with C-H distances in the range 0.95–0.99 Å and $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: *COLLECT* (Nonius, 2000); cell refinement: *DENZO–SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO–SMN* and *SCALEPACK* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2003).

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