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2-(4-Chlorophenyl)-4-(2-hydroxyethyl)-5-methyl-2,4-dihydro-3 <i>H</i> -1,2,4-triazol-3-one
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# **Structure Reports Online**

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# S. Thamotharan, <sup>a</sup> V. Parthasarathi, <sup>a\*</sup> Vinay Sunagar, <sup>b</sup> Bharati Badami <sup>b</sup> and Anthony Linden <sup>c</sup>

<sup>a</sup>Department of Physics, Bharathidasan University, Tiruchirappalli 620 024, India, <sup>b</sup>Post-Graduate Department of Studies in Chemistry, Karnatak University, Dharwad 580 003, India, and <sup>c</sup>Institute of Organic Chemistry, University of Zürich, Winterthurerstrasse 190, CH-8057 Zürich, Switzerland

Correspondence e-mail: vpsarati@yahoo.com

### **Key indicators**

Single-crystal X-ray study  $T=160~\mathrm{K}$  Mean  $\sigma(\mathrm{C-C})=0.003~\mathrm{\mathring{A}}$  R factor = 0.043 wR factor = 0.122 Data-to-parameter ratio = 16.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 2-(4-Chlorophenyl)-4-(2-hydroxyethyl)-5-methyl-2,4-dihydro-3*H*-1,2,4-triazol-3-one

In the title compound,  $C_{11}H_{12}ClN_3O_2$ , the dihedral angle between the phenyl and triazole rings is  $30.63~(9)^\circ$ . An intermolecular  $O-H\cdots O$  hydrogen bond is formed between the hydroxy group and the carbonyl group of the triazole moiety of a neighbouring molecule. This interaction links the molecules into chains, which run parallel to the c axis.  $C-H\cdots O$  intermolecular hydrogen bonds are also observed.

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### Comment

Extensive studies have been carried out on substituted 1,2,4-triazole derivatives (Cornelissen *et al.*, 1992; Kunkeler *et al.*, 1996; Chinnakali *et al.*, 1999; Fun *et al.*, 1999; Kumaran *et al.*, 1999). Research findings indicate that the 1,2,4-triazole moiety is associated with diverse pharmacological activities, such as analgesic, anti-asthmatic, diuretic, antifungal, antibacterial, pesticidal and anti-inflammatory activities (Bennur *et al.*, 1976; Heubach *et al.*, 1980; Sharma & Babel, 1982; Mohamed *et al.*, 1993). In view of this, the crystal structure determination of the title compound, (I), has been carried out in order to elucidate the stereochemistry and the molecular conformation.

$$C \vdash \bigvee_{N \longrightarrow N} O \\ CH_3 \longrightarrow OH$$

The bond lengths and angles in (I) are comparable with those reported for related structures (Chen et al., 1998; Wang et al., 1998). Unweighted least-squares planes calculations show that the phenyl group is oriented at an angle of 30.63 (9)° with respect to the plane of the triazole ring. The hydroxyethyl group projects roughly perpendicular to the triazole ring  $[C3-N4-C12-C13 = 80.7 (2)^{\circ}]$ . The exocyclic angle N2-C3-O3 [128.58 (17)°] is significantly larger than the normal value of 120°, and this may be due to the short contact between atoms H11 of the phenyl ring and O3 (2.54 Å). The hydroxy group forms an intermolecular hydrogen bond with the carbonyl O atom of an adjacent molecule. This interaction links the molecules into chains, which run parallel to the c axis and have a graph-set motif of C(7) (Bernstein et al., 1995). Several C-H···O intermolecular hydrogen bonds are also observed in (I) (Table 2).

## **Experimental**

The title compound was prepared by heating 3-(4-chlorophenyl)-5-methyl-2-oxo- $\Delta^4$ -1,3,4-oxadiazole with ethanolamine. The solid obtained, (I), was crystallized from ethanol (m.p. 388–403 K).

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# organic papers

## Crystal data

$C_{11}H_{12}ClN_3O_2$	$D_x = 1.492 \text{ Mg m}^{-3}$
$M_r = 253.69$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 13 330
a = 12.6968 (4)  Å	reflections
b = 11.1886 (3)  Å	$\theta = 2.0 - 27.5^{\circ}$
c = 7.9872 (2)  Å	$\mu = 0.33 \text{ mm}^{-1}$
$\beta = 95.446 \ (1)^{\circ}$	T = 160 (2)  K
$V = 1129.54 (5) \text{ Å}^3$	Tablet, pale yellow
Z = 4	$0.28 \times 0.20 \times 0.10 \text{ mm}$

### Data collection

Nonius KappaCCD diffractometer	1777 reflections with $I > 2\sigma(I)$
$\varphi$ and $\omega$ scans with $\kappa$ offsets	$R_{\rm int} = 0.081$
Absorption correction: multi-scan	$\theta_{\rm max} = 27.5^{\circ}$
(Blessing, 1995)	$h = -16 \rightarrow 16$
$T_{\min} = 0.895, T_{\max} = 0.970$	$k = -14 \rightarrow 14$
23 311 measured reflections	$l = -10 \rightarrow 10$
2597 independent reflections	

### Refinement

Refinement on $F^2$	H atoms treated by a mixture of
$R[F^2 > 2\sigma(F^2)] = 0.043$	independent and constrained
$wR(F^2) = 0.122$	refinement
S = 1.06	$w = 1/[\sigma^2(F_o^2) + (0.0683P)^2]$
2597 reflections	where $P = (F_o^2 + 2F_c^2)/3$
159 parameters	$(\Delta/\sigma)_{\rm max} < 0.001$
-	$\Delta \rho_{\text{max}} = 0.24 \text{ e Å}^{-3}$
	$\Delta \rho_{\min} = -0.37 \text{ e Å}^{-3}$

Table 1 Selected geometric parameters (Å, °).

N1-C5	1.296 (2)	N4-C5	1.377 (2)
N1-N2	1.401(2)	N4-C3	1.383 (2)
N2-C3	1.370 (2)		
C5-N1-N2	104.30 (15)	N2-C3-N4	103.37 (16)
C3-N2-N1	112.02 (15)	N1-C5-C15	124.96 (17)
C5-N4-C3	108.20 (16)		

Table 2 Hydrogen-bonding geometry (Å, °).

$D-H\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
$\begin{array}{c} O14-H14\cdots O3^{i} \\ C12-H122\cdots O14^{ii} \\ C12-H121\cdots O14^{iii} \\ C15-H153\cdots O14^{i} \end{array}$	0.86 (3)	1.91 (3)	2.7498 (19)	163 (3)
	0.99	2.57	3.528 (2)	163
	0.99	2.59	3.564 (2)	170
	0.98	2.49	3.463 (2)	174

Symmetry codes: (i)  $x, \frac{3}{2} - y, \frac{1}{2} + z$ ; (ii)  $x, \frac{3}{2} - y, z - \frac{1}{2}$ ; (iii)  $-x, y - \frac{1}{2}, \frac{1}{2} - z$ .

The position of the hydroxy H atom was determined from a difference Fourier map and refined freely along with its isotropic displacement parameter. The methyl H atoms were constrained to an ideal geometry (C-H = 0.98 Å), with  $U_{iso}(H) = 1.5U_{eq}(C)$ , but were allowed to rotate freely about the parent C-C bond. All remaining H atoms were placed in geometrically idealized positions (C-H =0.95-0.99 Å) and constrained to ride on their parent atoms with  $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm C}).$ 

Data collection: COLLECT (Nonius, 2000); cell refinement: DENZO-SMN (Otwinowski & Minor, 1997); data reduction:

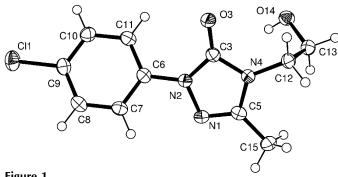


Figure 1

View of the asymmetric unit of the title compound, showing the atomlabelling scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are represented by circles of arbitrary radii.

DENZO-SMN and SCALEPACK (Otwinowski & Minor, 1997); program(s) used to solve structure: SIR92 (Altomare et al., 1994); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Version 1.07; Farrugia, 1997); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2002).

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